

MASA MOLEKULAR ERLATIBOA

SUBSTANTZIA	ATOMOA ALA MOLEKULA	ELEMENTUA ALA KONPOSATUA	MASA ERLATIBOAren KALKULOA
HCl	Molekula	konposatua	$ \begin{array}{cc} \text{H} & \text{Cl} \\ \downarrow & \downarrow \\ M_{\text{HCl}} = 1\text{u} + 35,5\text{u} = 36,5\text{u} \end{array} $
Ag	Atomoa	Elementua	$M_{\text{Ag}} = 107,87\text{u} \approx 107,9\text{u}$
H ₂ SO ₄	Molekula	konposatua	$M_{\text{H}_2\text{SO}_4} = 2 \cdot 1\text{u} + 32\text{u} + 4 \cdot 16\text{u} = 98\text{u}$
F	Atomoa	Elementua	$M_{\text{F}} = 19\text{u}$
F ₂ (g)	Molekula	elementua	$M_{\text{F}_2} = 38\text{u} \quad (19\text{u} \times 2)$
Fe ₃ (PO ₄) ₂	Molekula	konposatua	$M_{\text{Fe}_3(\text{PO}_4)_2} = \underbrace{3 \cdot 55,9\text{u}}_{\text{Fe}} + \underbrace{2 \cdot 31\text{u}}_{\text{P}} + \underbrace{8 \cdot 16\text{u}}_{\text{O}} = 357,7\text{u}$