

FORMULA	IZENA	IZENA	FORMULA
$\overset{+4}{\text{Sn}}\overset{-2}{\text{O}}_2$	Eztainu(IV)oxidoa Eztainu dioxidoa	Nitrogeno(V)oxidoa	$\overset{+5}{\text{N}}_2\overset{-2}{\text{O}}_5$
$\overset{+1}{\text{H}}_2\overset{-2}{\text{Te}}$	Hidrogeno teleruroa Azido telurhidrikoa	Silanoa	$\overset{-4}{\text{Si}}\overset{+1}{\text{H}}_4$
$\overset{+2}{\text{Mg}}(\overset{-1}{\text{OH}})_2$	Magnesio hidroxidoa Magnesio dihidroxidoa	Amonio hidroxidoa	$\overset{+1}{\text{N}}\overset{-1}{\text{H}}_4\overset{-1}{\text{O}}\text{H}$
$\overset{+2}{\text{Zn}}\overset{-1}{\text{Cl}}_2$	Zink kloruroa Zink dikloruroa	Burdin trifluoruroa	$\overset{+3}{\text{Fe}}\overset{-1}{\text{F}}_3$
$\overset{+1}{\text{Ag}}_2\overset{-2}{\text{S}}\overset{+6}{\text{O}}_4$	Zilar sulfatoa Dizilar tetraoxido sulfatoa	Azido periodikoa	$\overset{+1}{\text{H}}\overset{-2}{\text{I}}\overset{+7}{\text{O}}_4$
$\overset{+1}{\text{H}}_3\overset{-2}{\text{As}}\overset{+1}{\text{O}}_2$	Azido ortohipoartseniosoa. Trihidrogeno (dioxido artseniatoa)	Azido metafosforosoa	$\overset{+1}{\text{H}}\overset{-2}{\text{P}}\overset{+3}{\text{O}}_2$
$\overset{+2}{\text{Cd}}\overset{-2}{\text{Se}}\overset{+6}{\text{O}}_4$	Kadmio seleniatao Kadmio tetraoxido seleniatao	Sodio peroxidoa	$\overset{+1}{\text{Na}}_2\overset{-2}{\text{O}}_2$
$\overset{+2}{\text{Pt}}(\overset{-1}{\text{Mn}}\overset{+7}{\text{O}}_4)_2$	Platino(II)permanganatoa Platino bis(tetraoxido manganatoa)	Azido kromikoa	$\overset{+1}{\text{H}}_2\overset{-2}{\text{Cr}}\overset{+6}{\text{O}}_4$
$\overset{+3}{\text{Al}}(\overset{-1}{\text{Br}}\overset{+7}{\text{O}}_4)_3$	Aluminio perbromatoa Aluminio tris(tetraoxido bromatoa)	Berun(IV)hiposeleni toa	$\overset{+4}{\text{Pb}}(\overset{-2}{\text{Se}}\overset{+2}{\text{O}}_2)_2$
$\overset{+2}{\text{Be}}\overset{-2}{\text{O}}_2$	Berilio peroxidoa Berilio dioxidoa.	Merkurio(II) perkloratoa	$\overset{+2}{\text{Hg}}(\overset{-1}{\text{Cl}}\overset{+7}{\text{O}}_4)_2$